



Natural Logarithms

A logarithm is an exponent. We are most familiar with the base 10 number system:

$$10^0 = 1, \text{ and } \log 1 = 0$$

$$10^1 = 10, \text{ and } \log 10 = 1$$

$$10^2 = 100, \text{ and } \log 100 = 2$$

However, many phenomena in nature follow an exponential decrease or increase. Natural or base e logs are based on $e = 2.718281828423536\dots$, which is just a number like π . You will use the **[ln]** and **[e^x]** keys on your calculator.

Calculate the following:

$$e^1 = \underline{\hspace{2cm}}$$

$$\ln 465 = \underline{\hspace{2cm}}$$

$$e^0 = \underline{\hspace{2cm}}$$

$$e^{6.1420} = \underline{\hspace{2cm}}$$

$$e^{-1} = \underline{\hspace{2cm}}$$

$$\ln 2 = \underline{\hspace{2cm}}$$

$$\ln 2.7183 = \underline{\hspace{2cm}}$$

$$\ln \left(\frac{1}{2}\right) = \underline{\hspace{2cm}}$$

Sometimes we must solve an equation like $I = I_0 e^{-\mu x}$ for x . This equation says that if you start with some quantity I_0 , after a distance x the quantity will be I . You might also see something like $R = R_0 e^{-\lambda t}$ which says that the quantity R_0 is reduced to R after time t has passed. Both μ and λ are constants. Example: Solve $I = I_0 e^{-\mu x}$ for x when $I = \frac{1}{2} I_0$.

$$I = I_0 e^{-\mu x} \Rightarrow \frac{1}{2} I_0 = I_0 e^{-\mu x}$$

$$\frac{1}{2} = e^{-\mu x}$$

$$\ln\left(\frac{1}{2}\right) = \ln(e^{-\mu x}) \Rightarrow \ln\left(\frac{1}{2}\right) = -\mu x$$

$$x = \frac{\ln\left(\frac{1}{2}\right)}{-\mu} = \frac{\ln 2}{\mu} = \frac{.6931}{\mu}$$

A numerical example: Suppose $\mu = .01/\text{cm}$. Find x for $I = \frac{1}{2} I_0$. (This is referred to as the Half Value Length or Half Intensity Thickness.) We can use the results of the previous symbolic example.

$$x = \frac{.6931}{\mu}$$

$$x = \frac{.6931}{.001/\text{cm}}$$

$$x = 6.93 \text{ cm}$$

Now let's work a numerical answer where you use the [e^x] key. $R = R_0 e^{-\lambda t}$ describes the activity of a radioactive substance. Suppose $R_0 = 1000$ Curies, a common unit of activity. Suppose Iodine 131 (half life 8.04 days) is delivered to a thyroid patient via an IV. The decay constant, λ , is .0862/day. Find the activity R after two days have passed. Put the numbers in your calculator and make sure you can get the answer.

$$R = R_0 e^{-\lambda t}$$

$$R = (1000 \text{ Curies}) e^{-(.0862/\text{day})(2 \text{ days})}$$

$$R = (1000 \text{ Curies}) e^{-.1724}$$

$$R = (1000 \text{ Curies})(.84164)$$

$$R = 842 \text{ Curies}$$

Challenge Problem: How long would you need to wait for the ^{131}I activity to drop to only 5% of what it was originally? Hint: Let $R = .05R_0$.